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## Crystallization and preliminary X-ray diffraction studies of the BTL2 lipase from the extremophilic microorganism *Bacillus thermocatenulatus*

*Bacillus thermocatenulatus* lipase 2 (BTL2) is a thermoalkalophilic lipase that has been reported as an enantioselective biocatalyst for diverse reactions and that heads a group of enzymes that share high resistance towards many inactivation agents (heat, organic solvents, pH *etc.*). This makes BTL2 an important research target because of its potential industrial applications. BTL2 was cloned and overexpressed in *Escherichia coli*, purified and concentrated for crystallization using the sitting-drop vapour-diffusion method at 291 K. Crystals grew from a mixture of 13% MPD and 0.2 M ammonium acetate in 0.05 M sodium citrate pH 5.5–5.6. The crystals, which belonged to the orthorhombic space group *I*222 with unit-cell parameters  $a = 73.07$ ,  $b = 129.08$ ,  $c = 127.49$  Å, allowed the collection of an X-ray data set to 2.2 Å resolution.

### 1. Introduction

Lipases (triacylglycerol acylhydrolases; EC 3.1.1.3) are the most common enzymes used in biocatalysis (Ghanem & Aboul-Enein, 2004, 2005). These proteins have an extraordinary catalytic activity and substrate specificity over a wide range of conditions, including high organic solvent concentrations, high temperatures (food chemistry) or both (*e.g.* biodiesel production; Ghanem & Aboul-Enein, 2005; Aravindan *et al.*, 2007).

Molecular stability is a critical requirement for the suitability of an enzyme as a biocatalyst in all biotechnological applications (Burton *et al.*, 2002; Ó'Fágáin, 2003). Enzymes from thermophilic microorganisms are especially resistant towards common inactivation agents (heat, organic solvents, pH *etc.*) owing to their natural structural stability and rigidity (Owusu & Cowan, 1989; Cowan, 1997), which allows them to play an important role in several industrial applications (Schmidt-Dannert *et al.*, 1997). Moreover, their natural properties (specificity, selectivity, activity and stability) can be artificially improved further *via* immobilization techniques, therefore amplifying the range of conditions under which these enzymes can be used (Mateo *et al.*, 2007).

The thermoalkalophilic bacterial lipases show a high amino-acid sequence identity (95%) and share significant homology (30–35%) to lipases that are involved in the pathogenic processes of some infectious Gram-positive organisms such as *Staphylococcus* strains. The fact that thermophilic and *Staphylococcus* lipases (of unknown structure) are grouped into the I.5 family (Rosenstein & Götz, 2000) and share significant homology opens new possible approaches for drug development (enzyme inhibitors) against *Staphylococcus* species based on the structure of BTL2.

The lipase BTL2 (43 kDa) from *Bacillus thermocatenulatus* is an enzyme that has high stability towards both organic solvents and



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thermal conditions (Schmidt-Dannert *et al.*, 1994, 1997) and that has been reported (Palomo *et al.*, 2003) as an interesting enantioselective biocatalyst in diverse reactions, but its structure has not yet been determined. Here, we describe the initial results obtained for the crystallization of BTL2 and its X-ray diffraction data to 2.2 Å resolution.

## 2. Experimental

### 2.1. Expression and purification of BTL2

*Escherichia coli* BL21 (DE3) strain cells were transformed with the pT1BTL2 plasmid containing the *B. thermocatenuatus* gene that codes for the mature lipase BTL2, as described previously (Schmidt-Dannert *et al.*, 1996). Cells carrying the recombinant pT1BTL2 plasmid were grown at 303 K. Raising the temperature to 315 K for 20 h induced lipase overexpression.

The cells were collected by centrifugation, resuspended (10 ml per litre of culture) in buffer *A* (10 mM sodium phosphate pH 7.0, 3 mM benzamidine) and disrupted using a French press. The lysate was centrifuged at 5000g for 30 min at 277 K using a Sorvall centrifuge and the protein concentration was measured. The extract containing overexpressed BTL2 was diluted in buffer *A* to 5 mg ml<sup>-1</sup> protein. Octyl-Sepharose was then added [1:10(*w:v*)] using the batch method] to extract the protein and the mixture was gently stirred for 12 h at 298 K. The lipase adsorbed onto the octyl-Sepharose was eluted with 400 ml buffer *B* [12 mM sodium phosphate pH 7 and 0.125% (*v/v*) Triton X-100] and the supernatant (S1) was obtained by filtration. The BTL2 recovered in the S1 fraction was mixed with 25 g PEI-agarose and stirred gently for 2 h at 298 K (in this step, the BTL2 was not adsorbed by the resin). The support (PEI-agarose) with the adsorbed contaminating proteins was then discarded and the supernatant (S2) with the remaining BTL2 activity (near to 100%) was recovered by filtration. 10 g SP-550 Toyopearl was then added to the S2 supernatant using the batch method and the suspension was filtered; the new supernatant (S3) containing the BTL2 activity was collected. The supernatant (S3) temperature was decreased to 277 K and solid sodium bicarbonate was added to a final concentration of 12.5 mM. The pH was readjusted to 10.0 using NaOH. Q-Sepharose (10 g) was then added to this suspension also using the batch method and BTL2 was adsorbed onto this support. After a few minutes, the support (Q-Sepharose) was recovered by filtration and washed in a sintered glass funnel with distilled water containing 10 nM Triton X-100. The support was resuspended in 50 ml buffer *C* (500 mM sodium phosphate pH 7) to desorb the enzyme. The enzyme was concentrated 15-fold by centrifugation using an Amicon Ultra-15 membrane. Finally, the purified enzyme was dialyzed against double-distilled water.

### 2.2. Crystallization

High-throughput techniques with a NanoDrop robot (Innovadyne Technologies Inc.) were used to assay crystallization conditions using a few milligrams of pure BTL2 (5 mg ml<sup>-1</sup> in double-distilled water) with Crystal Screens I, II and Lite, Index Screen and SaltRx from Hampton Research and PACT Suite and JCSG+ Suite from Qiagen. Initial assays were carried out by the sitting-drop vapour-diffusion method at 291 K on Innovaplate SD-2 microplates (Innovadyne Technologies Inc.), mixing 250 nl protein solution with 250 nl precipitant solution and equilibrating against 80 µl well solution. BTL2 microcrystals grew under a condition containing 15% MPD in 0.05 M sodium citrate pH 5.6 buffer with 0.1 M ammonium acetate as an additive. This initial condition was optimized using sitting drops by

**Table 1**

Data-collection statistics for BTL2 crystals diffracting to 2.2 Å resolution.

Values in parentheses are for the highest resolution shell.

Crystal data	
Space group	<i>I</i> 222
Unit-cell parameters (Å)	
<i>a</i>	73.07
<i>b</i>	129.08
<i>c</i>	127.49
Data processing	
Temperature (K)	100
Wavelength (Å)	0.9792
Resolution (Å)	63.75–2.2 (2.36–2.2)
Unique data	37406
Redundancy	4.1 (4.5)
Data completeness (%)	99.8 (100)
Average <i>I</i> / $\sigma$ ( <i>I</i> )	8.9 (2.5)
Molecules per ASU	1
Matthews coefficient (Å <sup>3</sup> Da <sup>-1</sup> )	3.27
Solvent content (%)	62
<i>R</i> <sub>merge</sub> <sup>†</sup>	0.14 (0.48)

<sup>†</sup>  $R_{\text{merge}} = \frac{\sum_{hkl} \sum_i |I_i(hkl) - \langle I(hkl) \rangle|}{\sum_{hkl} \sum_i I_i(hkl)}$ , where  $I_i(hkl)$  is the *i*th measurement of reflection *hkl* and  $\langle I(hkl) \rangle$  is the weighted mean of all measurements.

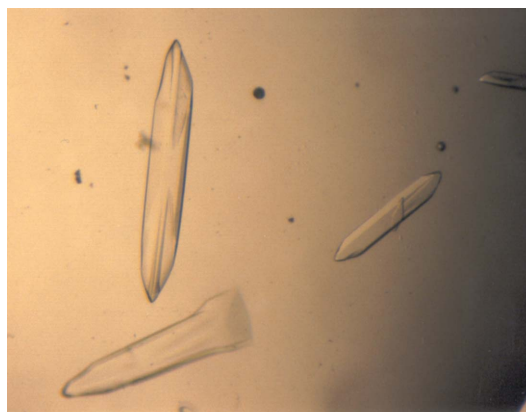
mixing 1 µl protein solution with 1 µl precipitant solution and equilibrating against 600 µl well solution. Good-quality crystals with a rice-grain shape were obtained using 0.05 M sodium citrate pH 5.6, 13% MPD and 0.2 M ammonium acetate. Crystals reached their maximum dimensions of 0.3 × 0.1 × 0.1 mm in 3 d (Fig. 1).

### 2.3. X-ray data collection and processing

Prior to flash-cooling to 100 K using a cryogenic system, all crystals were soaked for 5 s in a cryoprotectant solution consisting of 20% (*v/v*) glycerol in the crystallization solution. A native data set was collected using synchrotron radiation at ESRF (Grenoble) on beamline BM16 using a ADSC reverse- $\phi$  detector and a wavelength of 0.9792 Å. Collected images were processed and scaled using *MOSFLM* (Leslie, 1992) and *SCALA*, respectively (Collaborative Computational Project, Number 4, 1994).

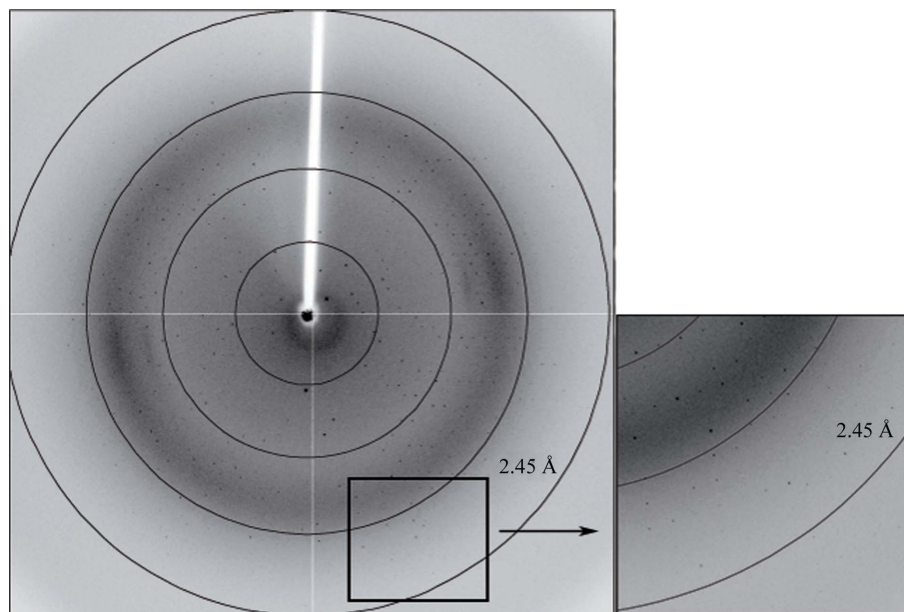
## 3. Results

An X-ray data set was collected to 2.2 Å resolution from a single BTL2 crystal and displayed apparently weak but good-quality diffraction patterns (Fig. 2). Despite some pattern weakness, preliminary X-ray data processing showed good processing statistics



**Figure 1**

BTL2 crystals obtained using 0.05 M sodium citrate pH 5.6, 13% MPD and 0.2 M ammonium acetate. The approximate dimensions of the crystals are 0.3 × 0.1 × 0.1 mm.



**Figure 2**  
(a) X-ray diffraction pattern of BTL2 crystals; (b) enlargement showing resolution rings.

(Table 1). The crystal belonged to the orthorhombic space group  $I222$ , with unit-cell parameters  $a = 73.07$ ,  $b = 129.08$ ,  $c = 127.49$  Å. Specific volume calculations based on the molecular weight of BTL2 and the unit-cell parameters indicated the presence of one monomer molecule in the asymmetric unit, with 63% solvent content ( $V_M = 3.27$  Å<sup>3</sup> Da<sup>-1</sup>; Matthews, 1968).

Structure determination was initiated using the lipase from *B. stearothermophilus* (PDB code 1ji3), which shows 95% sequence identity, as a structural model. Molecular replacement was performed with the *MOLREP* program (Vagin & Teplyakov, 1997) using reflections to 3.5 Å resolution. A single and unambiguous solution for the rotation and translation functions was obtained, which yielded a final correlation coefficient of 0.45 and an  $R$  factor of 0.46. The space group was confirmed to be  $I222$ , with one protein monomer in the asymmetric unit. Structural refinement of the BTL2 model is currently in progress.

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